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Key indicators

Single-crystal X-ray study
 $T = 299$ K
Mean $\sigma(\text{C}-\text{C}) = 0.008$ Å
 R factor = 0.094
 wR factor = 0.206
Data-to-parameter ratio = 16.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.*N*-(3,5-Dichlorophenyl)-2,2,2-trimethyl-
acetamideMolecules of the title compound, $\text{C}_{11}\text{H}_{13}\text{Cl}_2\text{NO}$, are linked
into a chain running along the *b*-axis direction. The structure
shows a close resemblance to those of related amides but with
slightly different bond parameters.Received 11 March 2007
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Comment

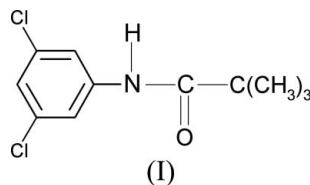
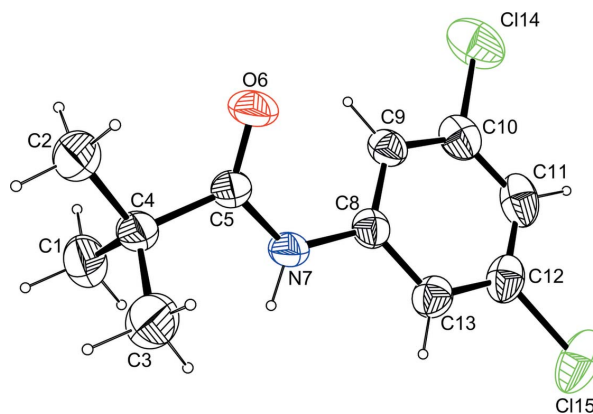
The amide unit is an important constituent of many biologi-
cally significant compounds. As part of a study of the effect of
ring and side-chain substituents on the solid-state geometry of
N-aromatic amides (Gowda *et al.*, 2006; Gowda, Kozisek *et al.*,
2007) we report here the crystal structure of *N*-(3,5-dichlorophenyl)-
2,2,2-trimethylacetamide, (I) (35DCPTMA) (Fig. 1). Comparison of the bond
parameters of 35DCPTMA with those of other amides, namely, *N*-(phenyl)-2,2,2-
trimethylacetamide (PTMA) and *N*-(3,5-dimethylphenyl)-
2,2,2-trimethylacetamide (35DMPTMA) (Gowda, Paulus *et al.*, 2007)
are shown in Table 2. Comparison revealed that *meta*-dichloro ring
substitution does not really affect the bond lengths, which are roughly
identical within experimental error.Molecules of the title compound are linked by $\text{N}-\text{H}\cdots\text{O}$
hydrogen bonds (Table 1), forming a chain running along the *b*
axis (Fig. 2).

Figure 1

The molecular structure of the title compound, showing the atom-
labelling scheme. Displacement ellipsoids are drawn at the 30%
probability level.

Experimental

The title compound was prepared according to the literature method of Shilpa & Gowda (2007). The purity of the compound was checked by determining its melting point. It was characterized by recording its infrared and NMR spectra (Shilpa & Gowda, 2007). Single crystals of the title compound were obtained by slow evaporation of an ethanol solution.

Crystal data

$C_{11}H_{13}Cl_2NO$ $V = 2479.7 (13) \text{ \AA}^3$
 $M_r = 246.12$ $Z = 8$
 Orthorhombic, $Pbca$ $Mo\ K\alpha$ radiation
 $a = 10.655 (2) \text{ \AA}$ $\mu = 0.50 \text{ mm}^{-1}$
 $b = 9.999 (3) \text{ \AA}$ $T = 299 (2) \text{ K}$
 $c = 23.275 (9) \text{ \AA}$ $0.40 \times 0.14 \times 0.03 \text{ mm}$

Data collection

Oxford Diffraction Xcalibur 14622 measured reflections
 diffractometer 2250 independent reflections
 Absorption correction: analytical 1165 reflections with $I > 2\sigma(I)$
 (*CrysAlis RED*; Oxford $R_{int} = 0.078$
 Diffraction, 2006)
 $T_{min} = 0.855, T_{max} = 0.976$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.094$ 139 parameters
 $wR(F^2) = 0.206$ H-atom parameters constrained
 $S = 1.14$ $\Delta\rho_{max} = 0.26 \text{ e \AA}^{-3}$
 2250 reflections $\Delta\rho_{min} = -0.19 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry ($\text{\AA}, \text{^\circ}$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N7-H7\cdots O6^i$	0.86	2.08	2.900 (5)	159

Symmetry code: (i) $-x + \frac{1}{2}, y + \frac{1}{2}, z$.

Table 2

Comparison of selected geometric parameters ($\text{\AA}, \text{^\circ}$) of the title compound with those in other related amides.

Parameter	35DCPTMA	PTMA	35DMPTMA
Space group	<i>Pbca</i>	<i>Pca2</i> ₁	<i>Pbca</i>
Z	8	4	8
$C(r)-C(r)_{mean}$	1.370 (6)	1.377 (4)	1.381 (4)
$C(r)-C(r)_{min}$	1.357 (6)	1.362 (5)	1.363 (5)
$C(r)-C(r)_{max}$	1.387 (6)	1.384 (3)	1.393 (4)
$C(r)-N$	1.414 (6)	1.420 (2)	1.417 (3)
$N-C(O)$	1.363 (5)	1.348 (3)	1.342 (3)
$C-O$	1.192 (5)	1.219 (2)	1.223 (3)
$C(O)-C(side)$	1.515 (6)	1.532 (2)	1.523 (4)
$C(2r)-C(1r)-C(6r)$	119.9 (4)	119.7 (2)	119.6 (3)
$C(2r)-C(1r)-N$	123.1 (4)	122.4 (2)	122.1 (3)
$C(6r)-C(1r)-N$	117.0 (4)	117.9 (2)	118.2 (3)
$C(1r)-N-C(O)$	126.9 (4)	126.8 (2)	127.2 (2)
$N-C(O)-C(side)$	115.4 (4)	116.1 (2)	116.8 (2)
$N-C(O)-O$	120.5 (4)	122.1 (2)	120.9 (3)
$O-C(O)-C(side)$	124.1 (4)	121.9 (2)	122.2 (2)

Note: $r =$ ring and side = side chain.

All H atoms were positioned geometrically and treated as riding with $C-H = 0.93 \text{ \AA}$ (CH aromatic) or 0.96 \AA (CH_3) and $N-H = 0.86 \text{ \AA}$, with $U_{iso}(H) = 1.2U_{eq}(CH \text{ or } NH)$ and $U_{iso}(H) = 1.5U_{eq}(CH_3)$.

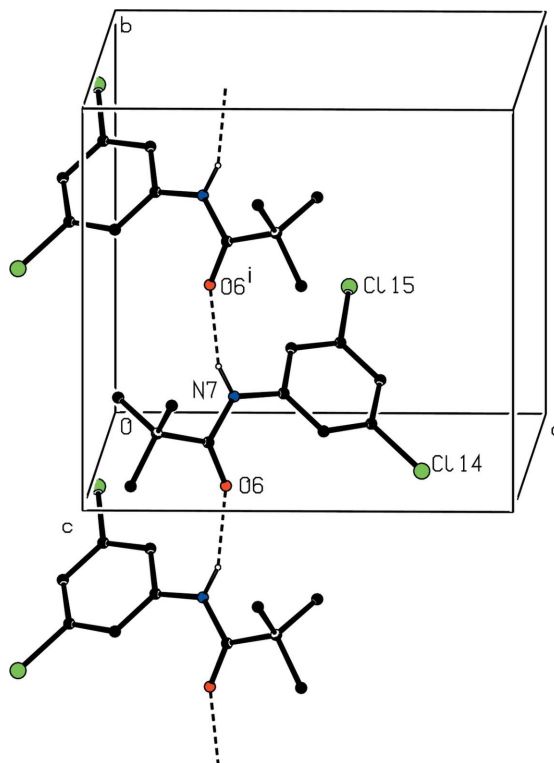


Figure 2

Partial packing view, showing the $N-H\cdots O$ hydrogen bonds linking the molecules into a chain. H bonds are shown as dashed lines. For the sake of clarity, H atoms not involved in hydrogen bonding have been omitted. [Symmetry code: (i) $\frac{1}{2} - x, \frac{1}{2} + y, z$]

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2006); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2006); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* (Sheldrick, 1997), *PLATON* and *WinGX* (Farrugia, 1999).

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